

β -K₂Cr₂O₇Sergey V. Krivovichev,^{a*} Elena V. Kir'yanova,^b Stanislav K. Filatov^b and Peter C. Burns^a^aDepartment of Civil Engineering and Geological Sciences, 156 Fitzpatrick, University of Notre Dame, Notre Dame, IN 46556-0767, USA, and ^bDepartment of Crystallography, St Petersburg State University, University Emb. 7/9, 199034 St Petersburg, Russia

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The monoclinic modification of dipotassium dichromate, β -K₂Cr₂O₇, has been synthesized in the K₂Cr₂O₇-H₂O system. The structure consists of K⁺ cations and Cr₂O₇²⁻ dimers. In contrast with triclinic α -K₂Cr₂O₇ [Kuz'min, Ilyukhin, Kharitonov & Belov (1969). *Krist. Tech.* **4**, 441–461], the Cr₂O₇²⁻ groups in β -K₂Cr₂O₇ have twofold crystallographic symmetry and are parallel to each other.

Comment

The monoclinic modification of K₂Cr₂O₇, *i.e.* β -K₂Cr₂O₇, was first described by Klement & Schwab (1960). Powder electron diffraction studies were performed by Zhukova & Pinsker (1964), who suggested space group *P*2₁/*c*. However, the structure resisted solution due to the absence of appropriate crystals. We present here the synthesis and single-crystal structure of β -K₂Cr₂O₇.

The crystal structure of β -K₂Cr₂O₇ is shown in Figs. 1(*a*) and 1(*b*). It consists of K⁺ cations and Cr₂O₇²⁻ dimers, which have twofold crystallographic symmetry and are composed of two CrO₄ tetrahedra sharing a common corner (atom O3; Fig. 2). The K atom is coordinated by eight O atoms forming a KO₈ square antiprism. The structure may be described as built from layers of composition [K₂Cr₂O₇] parallel to the (100) plane

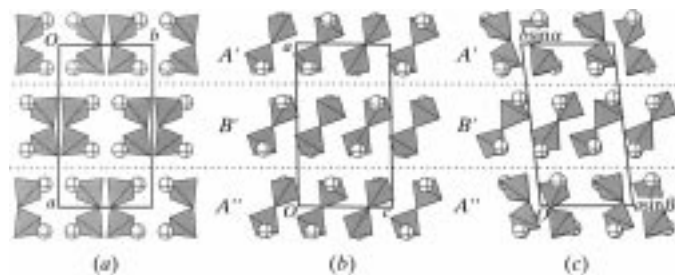


Figure 1

The crystal structures of (*a*) β -K₂Cr₂O₇ projected along the *c* axis, (*b*) β -K₂Cr₂O₇ projected along the *b* axis and (*c*) lopezite, α -K₂Cr₂O₇, projected along the *c* axis. Dashed lines indicate the layers (*A'*, *B'* and *A''*) composed of Cr₂O₇²⁻ anions (shown as polyhedra) and K⁺ cations (shown as spheres).

(Figs. 1*a* and 1*b*). All the Cr₂O₇²⁻ anions in β -K₂Cr₂O₇ are approximately parallel to each other.

Fig. 1 compares the crystal structure of β -K₂Cr₂O₇ with that of triclinic α -K₂Cr₂O₇ (lopezite; Kuz'min *et al.*, 1969). In contrast with β -K₂Cr₂O₇, the Cr₂O₇²⁻ groups in α -K₂Cr₂O₇ have two preferential positions at angles of 25–35° (Fig. 1*c*). The structure of α -K₂Cr₂O₇ may also be described as built from [K₂Cr₂O₇] layers [parallel with (010)]. The structure of the triclinic modification may be obtained from that of the monoclinic modification by rotation of one of the two adjacent layers by 90°. Thus, layers *A'* and *B'* shown in Fig. 1(*c*) have the same structure but are rotated relative to each other by 90° in comparison with the mutual position of adjacent layers (*A'*, *B'* and *A''*) in β -K₂Cr₂O₇.

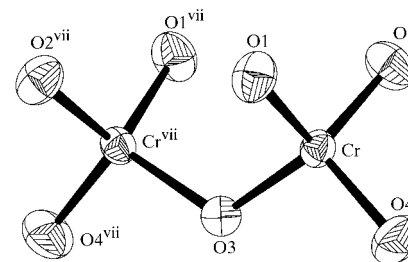


Figure 2

An ORTEP (Johnson, 1965) plot of the Cr₂O₇²⁻ anions in the title compound. Displacement ellipsoids are drawn at the 50% probability level. The symmetry code is as in Table 2.

Experimental

β -K₂Cr₂O₇ was synthesized in K₂CrO₄-K₂Cr₂O₇-H₂O polycomponent eutonic solutions (K₂O 19.17, CrO₃ 20.67, H₂O 60.16 wt%; saturation temperature 300.8 K, supercooling of 283–288 K).

Crystal data

K₂Cr₂O₇
M_r = 294.20
 Monoclinic, *C*2/*c*
a = 13.0339 (17) Å
b = 7.3750 (10) Å
c = 7.4672 (10) Å
 β = 91.923 (2)°
V = 717.38 (17) Å³
Z = 4

D_x = 2.724 Mg m⁻³
 Mo *K* α radiation
 Cell parameters from 523 reflections
 θ = 3–23°
 μ = 4.175 mm⁻¹
T = 293 (2) K
 Plate, orange
 0.46 × 0.24 × 0.09 mm

Data collection

Bruker PLATFORM diffractometer with a SMART CCD area detector
 ω scans
 Absorption correction: empirical via ψ scan (*SAINT-NT*; Bruker, 1998)
T_{min} = 0.317, *T_{max}* = 0.686

1779 measured reflections
 798 independent reflections
 718 reflections with *I* > 2 σ (*I*)
R_{int} = 0.045
 θ_{max} = 28.26°
h = -16 → 17
k = -9 → 9
l = -6 → 9

Refinement

Refinement on *F*²
R [*F*² > 2 σ (*F*²)] = 0.029
wR(*F*²) = 0.084
S = 1.107
 798 reflections
 52 parameters
 $w = 1/[\sigma^2(F_o^2) + (0.0513P)^2 + 0.5383P]$
 where $P = (F_o^2 + 2F_c^2)/3$

(Δ/σ)_{max} < 0.001
 $\Delta\rho_{max}$ = 0.50 e Å⁻³
 $\Delta\rho_{min}$ = -0.52 e Å⁻³
 Extinction correction: *SHELXTL* (Bruker, 1997)
 Extinction coefficient: 0.0074 (11)

Table 1

Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2).

$$U_{\text{eq}} = (1/3)\sum_i \sum_j U^{ij} a^i a^j \mathbf{a}_i \cdot \mathbf{a}_j.$$

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
Cr	0.39714 (3)	0.17429 (5)	0.13497 (5)	0.0213 (2)
K	0.14473 (5)	0.15065 (9)	0.35562 (8)	0.0328 (2)
O1	0.44153 (18)	0.3071 (3)	-0.0176 (3)	0.0337 (5)
O2	0.33753 (18)	0.2974 (3)	0.2780 (3)	0.0341 (5)
O3	1/2	0.0595 (4)	1/4	0.0283 (6)
O4	0.32025 (19)	0.0254 (3)	0.0494 (3)	0.0408 (6)

Table 2

Selected geometric parameters (\AA , $^\circ$).

Cr—O1	1.624 (2)	K—O1 ⁱⁱⁱ	3.039 (2)
Cr—O2	1.620 (2)	K—O2 ^{iv}	2.764 (2)
Cr—O3	1.7819 (13)	K—O2 ⁱⁱⁱ	2.802 (2)
Cr—O4	1.605 (2)	K—O2	2.814 (3)
K—O1 ⁱ	2.746 (2)	K—O4 ^v	2.886 (2)
K—O1 ⁱⁱ	2.859 (2)	K—O4 ^{vi}	2.965 (2)
O4—Cr—O2	109.82 (13)	O2—Cr—O3	108.56 (9)
O4—Cr—O1	111.46 (12)	O1—Cr—O3	110.17 (9)
O2—Cr—O1	108.33 (12)	Cr ^{vii} —O3—Cr	123.25 (15)
O4—Cr—O3	108.45 (11)		

Symmetry codes: (i) $\frac{1}{2} - x, \frac{1}{2} - y, -z$; (ii) $x - \frac{1}{2}, \frac{1}{2} - y, \frac{1}{2} + z$; (iii) $\frac{1}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$; (iv) $\frac{1}{2} - x, \frac{1}{2} - y, 1 - z$; (v) $\frac{1}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z$; (vi) $x, -y, \frac{1}{2} + z$; (vii) $1 - x, y, \frac{1}{2} - z$.

Data collection: *SMART-NT* (Bruker, 1998); cell refinement: *SAINTE-NT* (Bruker, 1998); data reduction: *SAINTE-NT*; program(s) used to solve structure: *SHELXTL* (Bruker, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *CRYSTAL-MAKER* (Palmer, 1999); software used to prepare material for publication: *SHELXTL*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FG1580). Services for accessing these data are described at the back of the journal.

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